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A novel cyclization to isoxazolo[3,4-e][2,1]benzisoxazole

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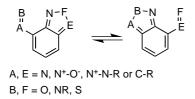
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Abstract—Methylation of 2,1-benzisoxazole 4,5-dione 4-oxime 2 using dimethyl sulfate in DMF and in the presence of potassium carbonate gave a substantial yield of isoxazolo[3,4-e][2,1]benzisoxazole 4 by an unexpected cyclization reaction of the *O*-methylation product 3. © 2002 Elsevier Science Ltd. All rights reserved.

The Boulton–Katritzky rearrangement (BKR) generalized in Scheme 1 has received considerable attention from both experimental and theoretical standpoints.^{1a,1b} The rearrangement occurs with a wide variety of N-, Oand S-containing heterocycles the prototype of which is 4-nitrobenzofuroxan 1 which rearranges to an identical molecule probably via the transition state shown in Scheme 2 according to calculations at the MP4(SDQ)/ $6-31G^*$ level.²

The common feature of all these rearrangements is a pivotal nitrogen atom that probably suffers minimal change in geometry during the molecular transition. Recently the possibility of using carbon as the pivotal atom in a BKR has been considered theoretically³ and it was concluded that with suitable substituents (e.g. an

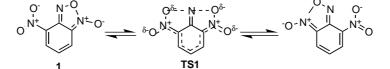


Scheme 1.

alkoxy or dialkylamino group) attached to the 5-position of 4-nitroso-2,1-benzisoxazole, such a rearrangement might be energetically possible through deprotonation of the isoxazole ring. During an experimental program designed to test the theoretical predictions we synthesized **2** obtained as the keto-oximino tautomer via the route shown in Scheme 3.

It was hoped that methylation of **2** would give 5methoxy-4-nitroso-2,1-benzisoxazole as a suitable precursor to attempt a base-catalyzed BKR but in fact the reaction gave a mixture of **3** (5%) and **4** (32%). The products were readily separated by chromatography on silica gel and characterized by MS, elemental analysis and ¹H/¹³C NMR.

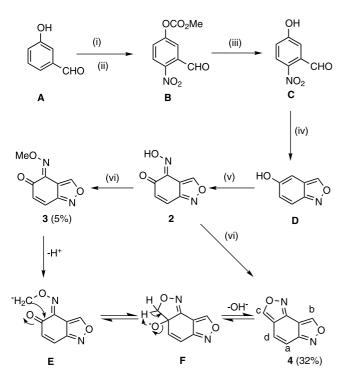
The cyclized product 4 was particularly intriguing since it appeared to be derived from 3 probably by the mechanism shown in Scheme 3. Convincing evidence to support this hypothesis was provided by heating a sample of pure 3 in DMF at 55° C for 1 h in the presence of anhydrous potassium carbonate. All the starting material was consumed and compound 4 was separated from the reaction mixture as the only isolable product. To the best of our knowledge there is no reported precedent for this cyclization. Apart from



Scheme 2.

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Scheme 3. (i) ClCO₂Me, pyridine; (ii) fuming HNO₃, conc. H_2SO_4 ; (iii) 10% aqueous NaOH; (iv) SnCl₂·H₂O, conc. HCl, 10–15°C; (v) NaNO₂, HCl, 0–5°C; (vi) Me₂SO₄, DMF, K₂CO₃, 55–60°C.

elemental analysis and HRMS, 4 was characterized by some unusual NMR features. In CDCl₃ just three proton signals with no fine structure were observed at 9.43, 8.19, and 7.65 ppm in a ratio of 1:1:2, respectively. In toluene- d_8 however, four signals were observed at 8.53 (d, J=1.05 Hz, 1H), 7.19 (s, 1H), 7.12 (dd, ${}^{3}J=9.6$ Hz, ${}^{4}J$ =1.05 Hz, 1H) and 6.80 ppm (d, J=9.6 Hz, 1H), which were assigned to protons c, b, d, and a, respectively. Selective homonuclear decoupling at 6.80 ppm collapsed proton b to a doublet $(J \sim 1 \text{ Hz})$ and selective decoupling at 7.12 ppm collapsed protons a and c to singlets consistent with the proposed structure. Presumably an intermolecular complex with the solvent toluene promoted separation of the ¹H NMR signals. The ¹³C NMR in CDCl₃ showed eight signals again consistent with 4 and ¹H-¹³C COSY correlation spectra enabled the final carbon assignments to be made.

In conclusion, although the original objectives of the project are still to be realized, a novel cyclization has been established involving formation of a carbon–carbon bond which may well afford a new route to derivatives of 2,1-benzisoxazoles through *o*-nitroso phenols or their keto-oxime tautomers.

Acknowledgements

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References

- (a) Boulton, A. J.; Ghosh, P. B.; Katritzky, A. R. Angew. Chem., Int. Ed. Engl. 1964, 3, 693; (b) Katritzky, A. R.; Gordeev, M. F. Heterocycles 1993, 35, 483.
- Eckert, F.; Rauhut, G. J. Am. Chem. Soc. 1998, 120, 13478.
- 3. Rauhut, G. J. Org. Chem. 2001, 66, 5444.
- 4. Skiles, J. W.; Cava, M. P. J. Org. Chem. 1979, 44, 409.
- Phillips, B. T.; Hartman, G. D. J. Heterocycl. Chem. 1986, 23, 897.
- 6. Vogel, A. *Textbook of Practical Organic Chemistry*, 4th ed.; Longman: London and New York, 1978; p. 677.
- 7. Methyl 3-formylphenyl carbonate, methyl 3-formyl-4nitrophenyl carbonate, **B**, and 2-nitro-5-hydroxybenzaldehyde, **C**, were prepared from 3-hydroxybenzaldehyde, **A**, using procedures described in Ref. 4. *Methyl* 3*formylphenyl carbonate* was obtained in 66% yield as colorless crystals, mp 47–49°C; ¹H NMR (300 MHz, CDCl₃): δ =3.94 (s, 3H), 7.46 (dd, J=8.1, 1.0 Hz, 1H, H-6), 7.57 (t, J=7.8 Hz, 1H, H-5), 7.72 (d, J=0.7 Hz, 1H, H-2), 7.79 (dd, J=7.7, 1.0 Hz, 1H, H-4), 10.01 (s, 1H); ¹³C NMR (75 MHz, CDCl₃) δ =55.6, 121.7, 127.1, 127.4, 130.2, 137.8, 151.6, 153.9, 190.9. Anal. Calcd for C₉H₈O₄: C, 60.00; H, 4.48. Found: C, 59.75; H, 4.49.

Methyl 3-formyl-4-nitrophenyl carbonate **B**, was obtained in 95% yield as purple prisms, mp 76–78°C; ¹H NMR (300 MHz, CDCl₃): δ =3.97 (s, 3H), 7.60 (dd, *J*=8.8, 2.6 Hz, 1H, H-6), 7.76 (d, *J*=2.6 Hz, 1H, H-2), 8.21 (d, *J*=8.8 Hz, 1H, H-5), 10.43 (s, 1H); ¹³C NMR (75 MHz, CDCl₃) δ , 56.1, 121.9, 125.7, 126.6, 133.2, 146.4, 152.8, 154.8, 186.9. Anal. Calcd for C₉H₇NO₆: C, 48.01; H, 3.13; N, 6.22. Found: C, 48.07; H, 3.07; N, 6.09.

2-Nitro-5-hydroxybenzaldehyde C, was obtained in 92% yield as yellow needles, mp 167–169°C, [lit. mp 167–168°C⁴]; ¹H NMR (300 MHz, DMSO- d_6): δ = 5.24 (brs, 1H), 6.92 (s, 1H, H-6), 6.97 (d, J=8.9 Hz, 1H, H-4), 8.07 (d, J=8.9 Hz, 1H, H-3), 10.28 (s, 1H); ¹³C NMR (75 MHz, DMSO- d_6) δ = 115.3, 119.6, 127.9, 135.7, 138.2, 166.5, 190.7.

5-Hydroxy-2,1-benzisoxazole D, was obtained in 90% yield as colorless prisms, mp 150-152°C by the procedure detailed in Ref. 5. ¹H NMR (300 MHz, DMSO- d_6): $\delta =$ 6.74 (s, 1H, H-4), 7.07 (d, J=9.5 Hz, 1H, H-6), 7.57 (d, J=9.5, Hz, 1H, H-7), 9.42 (s, 1H, H-3), 9.85 (s, 1H); ¹³C NMR (75 MHz, DMSO- d_6) $\delta = 96.0, 115.9, 118.5, 128.2,$ 153.0, 153.3, 153.4. Anal. Calcd for C₇H₅NO₂: C, 62.22; H, 3.73; N, 10.37. Found: C, 62.32; H, 3.73; N, 10.16. 2,1-Benzisoxazole-4,5-dione 4-oxime 2 was obtained in 100% yield as a grey-brown powder, mp 228-229°C (d) by the method outlined in Ref. 6. ¹H NMR (300 MHz, DMSO- d_6): $\delta = 6.64$ (d, J = 10.0 Hz, 1H, H-7), 7.89 (d, J=10.0 Hz, 1H, H-6), 9.70 (s, 1H, H-3), 14.12 (s, 1H, NOH); ¹³C NMR (75 MHz, DMSO- d_6) $\delta = 108.2$, 130.3, 134.8, 140.9, 154.1, 161.3, 181.7. Anal. Calcd for C₇H₄N₂O₃: C, 51.23; H, 2.46. Found: C, 51.58; H, 2.61. 2,1-Benzisoxazole-4,5-dione-4-(O-methyloxime) 3 and isoxazolo[3,4-e][2,1]benzisoxazole 4 were obtained as a mixture by methylation of 2 using the procedure outlined in Ref. 4. A mixture of 2 (1.0 g, 6.09 mmol), anhydrous potassium carbonate (1.57 g, 11.36 mmol), dimethyl sulfate (1.28 g, 10.18 mmol) and dry DMF (42 mL) was

heated with stirring at 54-60°C for 3 h under argon. The

mixture was cooled to 20°C and water (70 mL) was added slowly with stirring. The mixture was extracted with Et₂O (3×50 ml) and the combined ether extracts dried over anhydrous MgSO₄. The solvent was evaporated and the residue was purified by column chromatography (hexanes/ ethyl acetate, 6/1-3/1) to afford pure **3** as the first fraction (0.05 g, 5%) and **4** as the second fraction (0.31 g, 32%). 2,1-Benzisoxazole-4,5-dione-4-(O-methyloxime) **3**, colorless powder, mp 164–165°C; ¹H NMR (300 MHz, CDCl₃): δ =4.30 (s, 3H), 6.64 (d, J=10.4 Hz, 1H, H-7), 7.43 (d, J=10.4 Hz, 1H, H-6), 9.31 (s, 1H, H-3); ¹³C NMR (75 MHz, CDCl₃): δ =64.5, 109.3, 131.6, 139.5, 141.6, 153.1, 161.0, 178.2. LSIMS, Calcd for C₈H₇N₂O₃ (*M*+1): 179.0457. Found: 179.0451.

Isoxazolo[3,4-e][2,1]*benzisoxazole* 4, colorless powder, mp 155–157°C; ¹H NMR, (300 MHz, toluene- d_8): $\delta = 6.80$ (d, J = 9.6 Hz, 1H), 7.12 (dd, ³J = 9.6 Hz, ⁴J = 1.0 Hz, 1H) 7.19

(s, 1H), 8.53 (d, J=1.0 Hz, 1H), which were assigned to protons a, d, b and c, respectively (see Scheme 3). ¹³C NMR (75 MHz, CDCl₃): $\delta = 111.9$, 113.9, 118.0, 129.7, 146.8, 152.1, 152.3, 156.2. LSIMS, Calcd for C₈H₅N₂O₂ (M+1)=161.0351. Found: 161.0360. Anal. Calcd for C₈H₄N₂O₂: C, 60.01; H, 2.52; N, 17.49. Found: C, 59.61; H, 2.35; N, 17.10.

Conversion of 3 to 4

A mixture of 2,1-benzisoxazole-4,5-dione 4-(*O*-methyloxime) **3** (0.1 g, 0.56 mmol), anhydrous potassium carbonate (0.134 g, 0.94 mmol) and dry DMF (3 mL) was stirred at 54–57°C for 1 h under nitrogen. DMF was evaporated under reduced pressure. The residue was purified by column chromatography on silica gel using hexanes/ethyl acetate (6:1, v/v) as eluent to afford isoxazolo[3,4-e][2,1]benzisoxazole **4** (0.045 g, 0.28 mmol) as the only isolable product in 50% yield.